

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
21 April 2005 (21.04.2005)

PCT

(10) International Publication Number
WO 2005/035616 A1

(51) International Patent Classification⁷: **C08G 18/80**,
C09D 175/04

(21) International Application Number:
PCT/US2004/031108

(22) International Filing Date:
23 September 2004 (23.09.2004)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
10/677,789 2 October 2003 (02.10.2003) US

(71) Applicant (for all designated States except US):
TREMCO INCORPORATED [US/US]; 3735 Green
Road, Cleveland, OH 44122 (US).

(72) Inventor; and

(75) Inventor/Applicant (for US only): **FENG, Ta-Min**
[US/US]; 6775 Saint Ives Blvd., Hudson, OH 44236 (US).

(74) Agent: **DOCHERTY, Pamela, A.**; 800 Superior Avenue,
Suite 1400, Cleveland, OH 44114-2688 (US).

(81) Designated States (unless otherwise indicated, for every
kind of national protection available): AE, AG, AL, AM,
AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN,
CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,
GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE,
KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD,
MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG,
PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM,
TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM,
ZW.

(84) Designated States (unless otherwise indicated, for every
kind of regional protection available): ARIPO (BW, GH,
GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,
ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM),
European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI,
FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI,
SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ,
GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guid-
ance Notes on Codes and Abbreviations" appearing at the begin-
ning of each regular issue of the PCT Gazette.

(54) Title: PREPOLYMER COMPOSITIONS AND SEALANTS MADE THEREFROM

(57) **Abstract:** Prepolymer compositions for preparing a one-part, moisture-curable sealant, adhesive or coating, and a method of making such prepolymer compositions are provided. The prepolymer compositions comprise a polyurethane prepolymer that is end-capped with silane groups or a combination of silane groups and end capping groups that have been derived from an alcohol and an amount, referred to hereinafter as an "excess", of unreacted aromatic alcohol having a molecular weight of less than 2000. The excess of unreacted aromatic alcohol in the composition is from greater than 0% to 15% by mole of the level of original NCO groups in the prepolymer. From 50 to 100% of the original NCO groups are end-capped with silane groups, and from 0 to 50% of the original NCO groups are end-capped with a end-capping group that has been derived from an aromatic alcohol, an aliphatic alcohol or both. Sealant compositions and coating compositions comprising the present prepolymer composition are also provided.



WO 2005/035616 A1

PREPOLYMER COMPOSITIONS AND SEALANTS MADE THEREFROM

BACKGROUND

[0001] Sealants are used to provide liquid and gaseous barriers in various applications. Such applications include bonding of dissimilar materials, sealing of expansion joints, assembling curtain walls and side walls, weatherproofing, constructing roofing systems, and sealing the perimeters around doors, windows and other building components (i.e., perimeter sealing).

[0002] Sealant compositions can be of the one-part or two-part variety. Moisture curable, one-part sealant compositions generally contain end-capped polyurethane prepolymers and, usually, a curing catalyst that promotes a cross-linking reaction between the prepolymers when the sealant composition is exposed to atmospheric moisture. Upon application under normal conditions of temperature and moisture, one-part sealant compositions react to form tough, pliable elastomeric seals.

[0003] Sealant and coating compositions desirably have a combination of properties which render them particularly suitable for their intended applications. Such compositions should be able to be packaged in sealed containers or cartridges and stored for relatively long periods of time without objectionably "setting up" or hardening (as a result of cross-linking). When applied as a caulking sealant or coating composition, they should form a relatively tack-free surface soon after being applied and exposed to atmospheric moisture, and should cure without the formation of bubbles within an acceptable time period. Such compositions should adhere tenaciously in the

cured state to a wide variety of surfaces, such as to glass, aluminum, concrete, marble and steel surfaces. The sealant or coating in the cured state should have sufficient elasticity and flexibility to withstand expansions and contractions of panels, etc. with which it is associated during temperature variations that result from climatic changes, and to withstand wind forces that cause panels with which it is associated to flex or twist.

[0004] Polyurethane sealant and coating compositions typically are based on isocyanate-terminated prepolymers. In some cases, such prepolymers are fully or partially end-capped with silane groups. In general, sealants made with prepolymers that have been 100% end-capped with silane groups are not sufficiently flexible to sustain large joint movement. Accordingly, prepolymers that are end capped with a combination of silane groups and other groups, particularly aliphatic alcohol groups, have been developed. Although sealants made with prepolymers that have been end capped with a combination of silane and aliphatic alcohol groups have enhanced flexibility, these sealants are more likely to chalk, crack and yellow when exposed to extreme weathering conditions of high heat, high moisture, and prolonged exposure to UV radiation. Such sealants can also lose strength when exposed to these weathering conditions. Many of these undesirable changes can be attributed to reversion or degradation of the polymeric chains that are present in the cured sealant.

[0005] The polymeric chains of polyurethane prepolymers that have been partially end-capped with aliphatic alcohol end cappers can degrade during storage, particularly when exposed to high heat. As a result of such degradation sealant compositions made with these prepolymers lose their mechanical strength and are, thus, less desirable.

[0006] Accordingly, it is desirable to have new sealant compositions that provide elastomeric seals with improved weathering stability. New prepolymer compositions that have enhanced heat and storage stability are also desirable.

SUMMARY OF THE INVENTION

[0007] The present invention provides a prepolymer composition for preparing a one-part, moisture-curable sealant, adhesive or coating, and a method of making such prepolymer composition.

[0008] The prepolymer composition comprises a polyurethane prepolymer that is end-capped with silane groups or a combination of silane groups and end capping groups that have been derived from an alcohol and an amount, referred to hereinafter as an "excess", of aromatic alcohol. Preferably, the aromatic alcohol has a molecular weight of less than 2000. The silane end-capped prepolymer is formed by reacting a hydroxy-terminated polymeric material with an aromatic or, preferably, an aliphatic isocyanate to provide a prepolymer having terminal NCO groups, referred to hereinafter as the "original" NCO groups. The resulting polyurethane prepolymer is then partially or fully blocked or end-capped by reacting the prepolymer with a sufficient amount of a silane end capper to provide a polyurethane prepolymer in which 50% to 100% of the original NCO groups are capped with silane groups.

[0009] In those cases where the prepolymer is 100% end capped with silane groups, the prepolymer is then combined with a sufficient amount of the aromatic alcohol to provide a composition comprising an excess of unreacted aromatic alcohol and the prepolymer. In those cases where the prepolymer is partially end-capped with silane groups, the prepolymer may be reacted with aromatic alcohol to provide a composition comprising an excess of aromatic alcohol and a prepolymer that is fully end-capped with a combination of silane groups and aromatic alcohol groups. Alternatively, the prepolymer that is partially end-capped with silane groups may be reacted with an aliphatic alcohol in the presence of an aromatic alcohol to provide a composition comprising an excess of aromatic alcohol and a prepolymer that is end-capped with a combination of silane groups, aliphatic alcohol and aromatic alcohol groups. Alternatively, the prepolymer that is partially end-capped with silane groups may be reacted with an aliphatic alcohol in the absence of an aromatic alcohol to provide a prepolymer that is end-capped with a combination of silane groups, and aliphatic alcohol groups, and then combined with an aromatic alcohol to provide a composition comprising an excess of aromatic alcohol and the end-capped polyurethane prepolymer.

[0010] Depending upon the percentage of NCO groups that are end capped with silane groups, the level of aromatic alcohol that is combined with the silane-capped, polyurethane prepolymer is from greater than 0% to about 65% by mole of the level of the original NCO groups in the prepolymer. Thus, since the aromatic alcohol will react with NCO groups that are not end capped with silane groups to provide a polyurethane prepolymer that is end-capped with

both silane groups and aromatic alcohol groups, the excess of aromatic alcohol in the composition is from greater than 0% to 15% by mole of the level of original NCO groups in the prepolymer.

[0011] The present invention also provides sealant compositions and coating compositions that are made with the present prepolymer composition. The sealant composition and coating composition comprise the present prepolymer composition, and a catalyst for increasing the cure rate of the sealant composition or the coating composition. Preferably, the sealant composition further comprises a reinforcing filler. Optionally, the sealant composition comprises one or more of an adhesion promoter, a rheology modifier, a moisture scavenger and a UV stabilizer.

[0012] The present prepolymer composition, sealant composition, and coating composition have improved weathering stability, heat stability and UV stability as compared to compositions that do not contain excess aromatic alcohol. The sealants that are formed by moisture curing the present sealant composition have excellent mechanical properties, such as high elongation, low modulus, and excellent adhesion. Because they are solvent free, the prepolymer and sealant compositions of the present invention are also environmentally friendly.

DETAILED DESCRIPTION OF THE INVENTION

[0013] The present invention relates to prepolymer compositions and to sealant compositions and coating compositions comprising the present prepolymer compositions.

PREPOLYMER COMPOSITION

[0014] In one aspect, the present invention provides prepolymer compositions with improved storage capabilities. Such prepolymer compositions comprise an amount of unreacted aromatic alcohol, referred to herein as an excess, and a polyurethane prepolymer whose terminal ends are fully end-capped either with silane groups or with a combination of silane groups and end capping groups that have been derived from one or more aromatic alcohols, or one or more aliphatic alcohols, or a combination of one more aromatic alcohols and one or more aliphatic alcohols. The prepolymer is made by reacting a hydroxy-terminated polymeric material with an isocyanate to provide a prepolymer chain having NCO groups at the ends thereof. The resulting

polyurethane prepolymer is then reacted with a sufficient amount of a silane end-capper to provide a polyurethane prepolymer in which from 50 to 100%, preferably from 70 to 100%, more preferably from 85 to 100% of the NCO groups are blocked or end-capped with a silane group.

[0015] Suitable hydroxy-terminated polymeric materials for preparing the present silane end-capped, polyurethane prepolymer include, but are not limited to di, tri, and tetra functional polyols, including polyether polyols, polyester polyols, acrylic polyols, and polyols comprising two or more hydroxyl groups and a straight or branched chain hydrocarbon.

[0016] Suitable polyether diols and triols include polyethylene ether diols or triols, polypropylene ether diols or triols, polybutylene ether diols or triols, polytetramethylene ether diols or triols, and block copolymers of such diols and triols.

[0017] Suitable hydroxy-terminated polyesters include any hydroxy-terminated polyester prepared from poly-basic acids or anhydrides (for example, adipic acid and phthalic anhydride) and polyols in which the hydroxyl functionality of the polyester prepolymer is greater than 2, preferably over 2.3. Polylactone containing hydroxyl groups are also suitable for making the prepolymer, particularly polycaprolactone diol and triol.

[0018] Suitable acrylic polyols include hydroxyl-terminated polyacrylate. Acrylates include, but are not limited to, butylacrylate, methylacrylate, methylmethacrylate, ethyl acrylate, 2-ethylhexyl acrylate or the mixture of above. Suitable polyols comprising two or more hydroxyl groups and a straight or branched hydrocarbon chain include hydroxyl functionalized polybutadiene. Other suitable polyols include polycarbonates having hydroxyl groups.

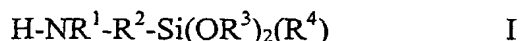
[0019] Preferably the polyol has a weight average molecular weight of from 500 to 18,000. For prepolymers that are used to make sealant compositions, it is preferred that the polyol have a weight average molecular weight of from 2,000 to 8,000. For prepolymers that are used to make coating compositions, it is preferred that the polyol have a weight average weight average molecular weight of from 500 to 4000.

[0020] The isocyanates that are reacted with the hydroxy-terminated backbone polymer are organic isocyanates having 2 or more isocyanate groups or a mixture of such organic

isocyanates. The isocyanates are aromatic or, preferably, aliphatic isocyanates. Examples of suitable aromatic di -or triisocyanates include p,p',p"-triisocyanato triphenyl methane, p,p'-diisocyanato diphenyl methane, naphthalene-1,5-diisocyanate, 2,4-tolylene diisocyanate, 2,6-tolylene diisocyanate, and mixtures thereof. Examples of preferred aliphatic isocyanates are isophorone diisocyanate, dicyclohexyl methane- 4,4' - diisocyanate, and mixtures thereof.

[0021] The polyurethane prepolymer may be prepared by mixing the hydroxy-terminated polymer and organic isocyanate together at ambient temperature and pressure, although the speed of the reaction is significantly increased if the temperature of the reaction mixture is raised to a higher temperature, for example, a temperature between 60°-100°C. A stoichiometric excess of the isocyanate is used to ensure that the polyurethane prepolymer has NCO terminal groups.

[0022] The resulting prepolymer having terminal NCO groups is then reacted with silane capping agents so that from 50 to 100%, preferably from 70 to 100%, more preferably from 85 to 100% by mole of the original terminal NCO groups are blocked or end-capped with silane groups. Examples of suitable silane capping agents include, but are not limited to, silanes corresponding to the formula I.



wherein R¹ represents hydrogen, a substituted aliphatic, cycloaliphatic, and/or aromatic hydrocarbon radical containing 1 to 10 carbon atoms, a second -R²-Si(OR³)₂(R⁴), or -CHR⁵-CHR⁶COOR⁷ where R⁵ and R⁶ are H or C₁₋₆ organic moiety, and R⁷ is C₁₋₁₀ organic moiety.

R² represents a linear or branched alkylene radical containing 1 to 8 carbon atoms.

R³ represents a C₁₋₆ alkyl group.

R⁴ = -CH₃, -CH₂CH₃, or OR³.

[0023] Examples of suitable aminosilanes corresponding to formula I include N-phenylaminopropyltrimethoxysilane, N-ethylaminoisobutyltrimethoxysilane, gamma-aminopropyltrimethoxysilane, gamma-aminopropyltriethoxysilane, and the reaction product of an aminosilane (such as gamma-aminopropyltrimethoxysilane, gamma-aminopropyltriethoxysilane, gamma-aminopropylmethyldimethoxysilane) with an acrylic monomer (such as ethyl acrylate, butyl acrylate, 2-ethylhexyl acrylate, methyl acrylate, methyl methacrylate, and glycidal acrylate).

[0024] Examples of other suitable silanes include mercaptosilane, the reaction product of a mercaptosilane with a monoepoxide, and the reaction product of an epoxysilane with a secondary amine.

[0025] Methods of preparing polyurethane prepolymers, and silylated polyurethane prepolymers are well known in the art. See, e.g., U.S. Pat. Nos. 3,627,722, 3,632,557, 3,979,344, and 4,222,925, which are incorporated herein by reference.

[0026] The resulting silylated, polyurethane prepolymers are then combined with a sufficient amount of an alcohol to block or end-cap any unreacted, i.e., non-silylated, NCO groups. The alcohol can be an aliphatic alcohol, or, preferably, an aromatic alcohol, or a combination of one or more aromatic alcohols and one or more aliphatic alcohols. The fully-blocked polyurethane prepolymer, i.e., the prepolymer whose NCO groups are 100% blocked with either silane groups or a combination of silane groups and alcohol groups is combined with a sufficient amount of an aromatic alcohol to provide a prepolymer composition comprising an end-capped polyurethane prepolymer and unreacted aromatic alcohol. Optionally, the composition may also comprise a small amount, i.e., less than 5 % by mole of original NCO groups of unreacted aliphatic alcohol. Depending upon the level of unreacted NCO groups, the silylated prepolymer is reacted with a level of alcohol, preferably an aromatic alcohol, which is greater than 0% to 65% by mole of the original NCO groups in the polyurethane prepolymer. Thus, the amount of unreacted or excess aromatic alcohol in the resulting composition is from greater than 0% to 15% by mole of the original NCO groups in the polyurethane prepolymer. Preferably, the amount of unreacted or excess aromatic alcohol in the resulting composition is greater than 2% by mole of the original NCO groups in the polyurethane prepolymer.

[0027] Preferably, the aromatic alcohol has a molecular weight less than 2000. Examples of suitable aromatic alcohols include phenol, 3-methoxyphenol, 4-methoxyphenol, nonylphenol, meta-cresol, para-cresol, 4-chlorophenol, meta-hydroxybenzaldehyde, ortho-hydroxybenzaldehyde, para-hydroxybenzaldehyde, hydroquinone, 1-hydroxy-2-propanone, 3-hydroxyacetophenone, and 4-hydroxyacetophenone.

[0028] The polyurethane prepolymer that is partially end-capped with silane groups is reacted with the aromatic alcohol at 80°C for 90 minutes to provide a composition comprising a

silane and aromatic alcohol end capped, polyurethane prepolymer and an excess of aromatic alcohol. Alternatively, the polyurethane prepolymer that is partially end-capped with silane groups is reacted with an aliphatic alcohol or a combination of an aliphatic and an aromatic alcohol at 80°C for 90 minutes to provide a composition comprising an polyurethane composition that is fully end-capped with a combination of silane groups and alcohol groups. In those instances where the prepolymer is 100% end-capped with silane groups or a combination of silane groups and alcohol end capping groups, the polymer may be combined with the excess of aromatic alcohol without heating.

[0029] Optionally, the composition also comprises a moisture scavenger which may be added after the reaction. Examples of moisture scavengers for inclusion in the prepolymer composition are vinytrimethoxysilane, methyltrimethoxysilane, hexamethyldisilazane, paratoluene sulfonyl isocyanate (PTSI), toluene diisocyanate (TDI), diphenyl methane diisocyanate (MDI), and polymeric MDI. The moisture content of the prepolymer preferably is 0.1 to 10%, more preferably 0.2 to 3%.

SEALANT AND COATING COMPOSITIONS

[0030] In another aspect, the present invention provides a sealant composition or coating composition which comprises the present prepolymer composition and a curing catalyst. Suitable curing catalysts include dibutyltin diacetate, dibutyltin dilaurate, and dibutyltin bis(acetylacetonate).

[0031] Preferably, the sealant composition or coating composition comprises a reinforcing filler. Examples of suitable reinforcing fillers include, but are not limited to, CaCO₃, fume silica, clay, talc. Preferably, the sealant composition also comprises an adhesion promoter. Suitable adhesion promoters include, but are not limited to, aminosilane and epoxysilane. Optionally, the sealant composition comprises one or more of, a rheology modifier, a moisture scavenger and a UV stabilizer. Examples of suitable moisture scavengers in sealant composition are, toluene diisocyanate (TDI), diphenyl methane diisocyanate (MDI), and polymeric MDI, vinyltrimethoxysilane, methyltrimethoxysilane, hexamethyldisilazane, paratoluene sulfonyl isocyanate

[0032] The compositions are prepared by mixing the prepolymer composition with the reinforcing filler until a homogenous mixture is achieved. The mixing is conducted under conditions which reduce the levels of moisture in the mixture. Thereafter, the catalyst and other desirable components, e.g., an adhesion promoter, are added to the mixture.

APPLICATION OF THE SEALANT AND COATING COMPOSITIONS

[0033] The sealant composition may be used to seal or bond buildings, highways, bridges, trucks, trailers, buses, recreational vehicles, cars, utilities, window applications, etc. Dissimilar materials that may, for example, be sealed or bonded with the present sealant composition include cement containing products, metals, plastics, glass, EIFS materials, vinyls, painted metals, coated glasses, Kynar, marble, granite, and brick. The sealant composition may be applied using standard equipment, such as a caulking gun or pump.

[0034] The coating composition may be applied to a surface by spraying or poured onto the surface and spread across the surface with a roller or other similar device.

EXAMPLES

[0035] The following examples are for purposes of illustration only and are not intended to limit the scope of the invention as defined in the claims which are appended hereto.

Example 1: Prepolymer Composition

[0036] 782 gram of polypropylene glycol diol having an average weight average molecular weight of 4000, 118 gram of diisodecyl phthalate (DIDP) and 51 gram of toluene diisocyanate (TDI) were reacted with stirring under nitrogen with 0.04 gram of dibutyltindilaurate at a temperature of 75 to 80°C. After the theoretical NCO content of about 0.7% was reached, 43 gram of an aminosilane (a reaction product of aminopropyltrimethoxysilane and an ethyl acrylate) was added to the reactor. The reaction continues until the free NCO content is below 0.05%. 3 gram of nonylphenol was then added to the reaction mixture along with 0.02 gram of dibutyltin dilaurate. Reaction continues at 75°C until % of NCO drops to zero. 3.3 gram of vinyltrimethoxysilane (a moisture scavenger) was added at the end of reaction. The resulting prepolymer composition had a Brookfield viscosity

of about 30,000 cps at 25°C. The Brookfield viscosity of the prepolymer composition is below 40,000cps after one year shelf aging.

Example 2: Prepolymer Composition

[0037] 868 gram of polypropylene glycol diol having an average weight average molecular weight of 4000 and 74.4 gram of isophorone diisocyanate were reacted under nitrogen with 0.06 gram of dibutyltin dilaurate at a temperature of 85°C. After the theoretical NCO content of about 0.85% was reached, 43 gram of an aminosilane (a reaction product of aminopropyltrimethoxysilane and an ethyl acrylate) was added to the reactor. The reaction continues until the free NCO content is below 0.2%. 11 gram of nonylphenol was then added to the reaction mixture. Reaction continues at 85°C until % of NCO drops to zero. 4 gram of vinyltrimethoxysilane (a moisture scavenger) was added at the end of reaction. The resulting prepolymer composition had a Brookfield viscosity of about 32,000 cps at 25°C. The Brookfield viscosity of the prepolymer composition is below 40,000 cps after one year shelf aging.

Example 3: Prepolymer Composition

[0038] 876 gram of polypropylene glycol diol having an average weight average molecular weight of 4000 and 75.4 gram of isophorone diisocyanate were reacted under nitrogen with 0.06 gram of dibutyltin dilaurate at a temperature of 85°C. After the theoretical NCO content of about 0.85% was reached, 34 gram of N-ethylaminoisobutyltrimethoxysilane was added to the reactor. The reaction continues until the free NCO content is below 0.2%. 11 gram of nonylphenol was then added to the reaction mixture. Reaction continues at 85°C until % of NCO drops to zero. 4 gram of vinyltrimethoxysilane (a moisture scavenger) was added at the end of reaction. The resulting prepolymer composition has a Brookfield viscosity of about 38,000 cps at 25°C. The Brookfield viscosity of the prepolymer composition is below 45,000 cps after one year shelf aging.

Example 4: Sealant compositions

[0039] Sealant compositions were produced based on the following formulation.

Materials	% by weight
Silane and nonylphenol endcapped	27.04

Polyurethane prepolymers (Example 1, 2, or 3)	
Plasticizer (Santicizer 160)	17
Coated precipitated calcium carbonate (Hakuenka CCR)	40
Coated ground calcium carbonate (Hi-Pflex 100)	10
Colorant	4.3
Vinyltrimethoxysilane (A-171)	1
Aminopropyltrimethoxysilane	0.08
N-ethylaminopropylisobutyltrimethoxysilane	0.5
Dibutyltindiacetate	0.08

[0040] The sealant compositions were prepared first by mixing the polyurethane prepolymer composition of Example 1, 2, or 3 with plasticizer, calcium carbonate and colorant at 75°C under vacuum for about 60 minutes to reduce the level of moisture in the system. All silanes and catalyst were then charged to the mixture, and mixing under vacuum continued for 10 minutes.

[0041] Each of the three sealant compositions prepared as described exhibited excellent stability upon being stored in sealed container at ambient condition for more than one year. Cured sealant developed a Shore A hardness of 20 after being exposed for three weeks at 50% humidity and 25°C. No cracks or chalks or any other indicators of reversion, i.e., degradation of the polymeric bonds, were observed in the sealant upon being exposed to UV light in Xenon Wertherometer for more than 4000 hours. Shore A hardness was still maintained at about 20 after 4000 hours exposure to UV light. In addition, sealant made with prepolymers of examples 2 and 3 gave excellent color stability after UV light exposure.

[0042] All sealants also provided excellent physical properties. Skin formation time for all sealants is about 60 minutes. The extrusion rate of the sealant is about 20 second for 20 gram of material at 60psi. Tensile strength at break was about 150 psi, elongation at break was more

than 500%, tear strength was about 40 pli. All sealants also gave excellent adhesion to various substrates, such as aluminum, glass, concrete, vinyl, steel, and brick.

[0043] Thus, the sealant or coating compositions of the present invention have excellent shelf-life, have excellent flowability for easy gunning, have good flexibility to withstand joint movement, do not shrink and gas upon curing, have excellent weathering stability upon exposure to UV light, and give fast skin time after exposed to the atmosphere moisture.

CLAIMS

What is claimed is:

1. A prepolymer composition for preparing sealant and coating compositions, comprising
 - a) prepolymers comprising terminal NCO groups that are end-capped with silane groups or with silane groups and alcohol groups, and
 - b) an excess of unreacted aromatic alcohol.
2. The prepolymer composition of claim 1 wherein the unreacted alcohol has a molecular weight of less than 2000.
3. The prepolymer composition of claim 1 wherein the excess of unreacted aromatic alcohol in the composition is from greater than 0% to 15% by mole of the end-capped NCO groups in the prepolymer.
4. The prepolymer composition of claim 3 wherein the excess of unreacted aromatic alcohol in the composition is greater than 2% by mole of the end-capped NCO groups in the prepolymer.
5. The prepolymer composition of claim 1,

wherein the composition further comprises an excess of unreacted aliphatic alcohol, wherein the excess of unreacted aliphatic alcohol in the composition is less than the excess of aromatic alcohol in the composition, and

wherein the excess of unreacted aliphatic alcohol in the composition is less than 5% by mole of the end-capped NCO groups in the prepolymer.
6. The prepolymer composition of claim 1 wherein from 50% to 100% of the NCO groups are end-capped with silane end-capping groups and wherein from 0% to 50% of the NCO groups on the prepolymer are end-capped with aromatic alcohol end-capping groups, or aliphatic alcohol end-capping groups, or a combination of aromatic and aliphatic alcohol end-capping groups.

7. The prepolymer composition of claim 1 wherein from 70% to 100% of the NCO groups on the prepolymer are end-capped with silane end-capping groups and wherein from 0% to 30% of the NCO groups on the prepolymer are end-capped with end-capped with aromatic alcohol end-capping groups, or aliphatic alcohol end-capping groups, or a combination of aromatic and aliphatic alcohol end-capping groups.

8. The prepolymer composition of claim 1 wherein from 80% to 100% of the NCO groups on the prepolymer are end-capped with silane end-capping groups and wherein from 0% to 20% of the NCO groups on the prepolymer are end-capped with end-capped with aromatic alcohol end-capping groups, or aliphatic alcohol end-capping groups, or a combination of aromatic and aliphatic alcohol end-capping groups.

9. The prepolymer composition of claim 1, wherein said composition further comprises a moisture scavenger.

10. The prepolymer composition of claim 9 wherein the moisture scavenger is selected from the group consisting of vinyltrimethoxysilane, methyltrimethoxysilane, hexamethyldisilazane, paratoluene sulfonyl isocyanate (PTSI), toluene diisocyanate (TDI), diphenyl methane diisocyanate (MDI), and polymeric MDI.

11. A prepolymer composition for preparing sealant and coating compositions, comprising

a) prepolymers comprising terminal NCO groups that are end-capped with silane groups or a combination of silane groups and aromatic alcohol groups, and

b) an excess of an aromatic alcohol having a weight average molecular weight of less than 2000.

wherein from 50% to 100% of the NCO groups on the prepolymers are end-capped with silane end-capping groups and wherein from 0% to 50% of the NCO groups on the prepolymers are end-capped with aromatic alcohol end-capping groups, and

wherein the excess aromatic alcohol in the composition is from greater than 0% to 15% by mole of the end-capped NCO groups in the prepolymer.

12. A method of preparing a prepolymer composition;

a) reacting a hydroxyterminated polymer with an organic isocyanate having 2 or more isocyanate groups to provide a prepolymer comprising terminal NCO groups;

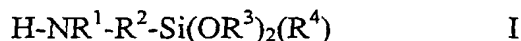
b) reacting the prepolymers of step (a) with silane capping agent to provide a prepolymer in which from 50% to 100% of the NCO groups on the prepolymer are end capped with a silane group;

c) optionally reacting the partially silane end-capped prepolymer of step (b) with an aromatic alcohol, an aliphatic alcohol, or a combination of an aromatic alcohol and an aliphatic alcohol to provide a polyurethane prepolymer that is partially end-capped with silane groups and partially end capped with alcohol groups.

d) reacting the prepolymer of step (b) or step (c) with a sufficient amount of an aromatic alcohol to provide a prepolymer composition comprising a fully end-capped NCO groups and an excess of aromatic alcohol.

13. The method of claim 12 wherein step d comprises reacting the prepolymer that is partially or fully end capped with silane group with a level of alcohol, said level being from greater than 0% to 65% by mole of the level of original NCO groups in the prepolymer.

14. The method of claim 12 wherein the silane capping agent has the following formula:



wherein R^1 represents hydrogen, a substituted aliphatic, cycloaliphatic, and/or aromatic hydrocarbon radical containing 1 to 10 carbon atoms, a second $\text{-R}^2\text{-Si(OR}^3\text{)}_2\text{(R}^4\text{)}$, or $\text{-CHR}^5\text{-CHR}^6\text{COOR}^7$ where R^5 and R^6 are H or C_{1-6} organic moiety, and R^7 is C_{1-10} organic moiety,

R^2 represents a linear or branched alkylene radical containing 1 to 8 carbon atoms,

R^3 represents a C_{1-6} alkyl group, and

$\text{R}^4 = \text{-CH}_3, \text{-CH}_2\text{CH}_3, \text{ or OR}^3$.

15. The method of claim 12 wherein the silane capping agent is selected from the group consisting of N-phenylaminopropyltrimethoxysilane, N-ethylaminoisobutyltrimethoxysilane, gamma-aminopropyltrimethoxysilane, gamma-aminopropyltriethoxysilane, the reaction product of an aminosilane with an acrylaic monomer, mercaptosilane, the reaction product of a mercaptosilane with a monoepoxide, and the reaction product of an epoxysilane with a secondary amine.
16. The method of claim 12 wherein the hydroxy-terminated polymer has a weight average molecular weight of from 500 to 18000.
17. The method of claim 12 wherein the aromatic alcohol has a molecular weight of less than 2000.
18. The method of claim 12 wherein the aromatic alcohol is selected from the group consisting of phenol, 3-methoxyphenol, 4-methoxyphenol, nonylphenol, meta-cresol, para-cresol, 4-chlorophenol, meta-hydroxybenzaldehyde, ortho-hydroxybenzaldehyde, para-hydroxybenzaldehyde, hydroquinone, 1-hydroxy-2-propanone, 3-hydroxyacetophenone, and 4-hydroxyacetophenone.
19. The method of claim 12 wherein the amount of aromatic alcohol that is reacted with the silane end-capped prepolymer is from greater than 0% to 65% by mole of the original NCO groups in the prepolymer.
20. The method of claim 8 wherein from 85% to 100% of the NCO groups in the prepolymer are end-capped with silane groups, and
- wherein the amount of aromatic alcohol that is reacted with the silane end-capped polyurethane prepolymer is from greater than 0% to 30% by mole of the original NCO groups in the prepolymer
21. The method of claim 12 wherein the hydroxy-terminated polymer is reacted with an aliphatic isocyanate.

22. The method of claim 12 wherein the aliphatic isocyanate is isophorone diisocyanate or dicyclohexyl methane- 4,4'- diisocyanate, or a mixture of isophorone diisocyanate and dicyclohexyl methane- 4,4'- diisocyanate.

23. The method of claim 12 wherein the hydroxy-terminated polymer is selected from the group consisting of a polyether polyol, a polyester polyol, an acrylic polyol, and a hydrocarbon chain comprising 2 or more hydroxyl groups.

24. A sealant composition comprising

a) a prepolymer composition comprising

i) prepolymers comprising terminal NCO groups that are end-capped with silane groups or a combination of silane groups and alcohol-derived end capping groups, wherein said prepolymers have a weight average molecular weight of from 5,000 to 60,000, and

ii) an excess of an aromatic alcohol , and

b) a catalyst for promoting cross-linking between the prepolymers when the sealant composition is exposed to atmospheric moisture.

25. The sealant composition of claim 24 wherein the excess of unreacted aromatic alcohol in the composition is from greater than 0% to 15% by mole of the end-capped NCO groups in the prepolymer

26. The sealant composition of claim 24 wherein from 50% to 100% of the NCO groups on the prepolymer are end-capped with silane end-capping groups and wherein from 0% to 50% of the NCO groups on the prepolymer are end-capped with aromatic alcohol end-capping groups, or aliphatic alcohol end-capping groups, or a combination of aromatic and aliphatic alcohol end-capping groups.

27. The sealant composition of claim 24 wherein from 70% to 100% of the NCO groups on the prepolymer are end-capped with silane end-capping groups and wherein from 0% to 30% of the NCO groups on the prepolymer are end-capped with with aromatic alcohol end-capping

groups, or aliphatic alcohol end-capping groups, or a combination of aromatic and aliphatic alcohol end-capping groups.

28. The sealant composition of claim 24 wherein from 80% to 100% of the NCO groups on the prepolymer are end-capped with silane end-capping groups and wherein from 0% to 20% of the NCO groups on the prepolymer are end-capped with aromatic alcohol end-capping groups, or aliphatic alcohol end-capping groups, or a combination of aromatic and aliphatic alcohol end-capping groups.

29. The sealant composition of claim 24 further comprising a reinforcing filler.

30. The sealant composition of claim 24 further comprising a moisture scavenger.

31. The sealant composition of claim 24 further comprising an adhesion promoter.

32. A coating composition comprising

a) a prepolymer composition comprising

i) prepolymers comprising terminal NCO groups that are end-capped with silane groups or a combination of silane groups and alcohol-derived end capping groups, wherein said prepolymers have a weight average molecular weight of from 1,000 to 20,000, and

ii) an excess of an aromatic alcohol, and

b) a catalyst for promoting cross-linking between the prepolymers when the coating composition is exposed to atmospheric moisture.

INTERNATIONAL SEARCH REPORT

International Application No

/US2004/031108

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 C08G18/80 C09D175/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08G C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	EP 0 676 403 A (OSI SPECIALTIES INC) 11 October 1995 (1995-10-11) page 3, line 21 - line 28; claims; examples	1-32
Y	EP 0 350 890 A (TREMCO INC) 17 January 1990 (1990-01-17) page 2, line 29 - line 45 page 4, line 10 - line 14 claims; examples	1-32
Y	US 4 507 443 A (BARRON LARRY R ET AL) 26 March 1985 (1985-03-26) column 4, line 19 - line 53 column 5, line 12 - line 22 examples	1-32
	----- -/--	



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

8 December 2004

Date of mailing of the international search report

15/12/2004

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Vaccaro, E

INTERNATIONAL SEARCH REPORT

International Application No

/US2004/031108

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 343 523 A (TREMCO INC) 29 November 1989 (1989-11-29) page 3, line 10 - line 26 page 4, line 5 - line 19 claims; examples -----	1-32

INTERNATIONAL SEARCH REPORT

tional Application No
/US2004/031108

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0676403	A	11-10-1995	DE 69512120 D1	21-10-1999
			DE 69512120 T2	03-02-2000
			EP 0676403 A1	11-10-1995
			JP 2594024 B2	26-03-1997
			JP 8053528 A	27-02-1996
EP 0350890	A	17-01-1990	US 4916199 A	10-04-1990
			AU 617837 B2	05-12-1991
			AU 3812189 A	18-01-1990
			DE 68911047 D1	13-01-1994
			DE 68911047 T2	14-04-1994
			EP 0350890 A2	17-01-1990
			ES 2061814 T3	16-12-1994
			JP 2086614 A	27-03-1990
			KR 162486 B1	15-01-1999
US 4507443	A	26-03-1985	AU 574757 B2	14-07-1988
			AU 3847885 A	29-08-1985
			CA 1244988 A1	15-11-1988
			DE 3562266 D1	26-05-1988
			EP 0156165 A1	02-10-1985
			HK 71390 A	21-09-1990
			JP 60195177 A	03-10-1985
EP 0343523	A	29-11-1989	US 4847319 A	11-07-1989
			AU 3494589 A	23-11-1989
			EP 0343523 A2	29-11-1989